
General Chemistry Principles And Modern Applications

Ubc Ed Vol 2

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principles of general chemistry - florida state college - course is an introduction to the concepts of inorganic chemistry including structures of matter, atomic theory, nomenclature, bonding, acids, bases and introduction to organic chemistry. this course is for students who have had no previous chemistry and plan to major in dental hygiene, medical technology, nursing or health related fields.

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a review of general chemistry - john wiley & sons - 2 chapter 1 a review of general chemistry must recognize that reactions occur as a result of the motion of electrons. example, in the following reaction the curved arrows represent the motion, or flow, of electrons. this flow of electrons causes the chemical change shown:

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1. prevention it is better to prevent waste than to treat or clean up waste after it has been created.
2. atom economy synthetic methods should be designed to maximize the incorporation of all materials used in the process into the final product.
3. less hazardous chemical syntheses

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