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homepage | wiley** - 2 chapter 1 a review of general chemistry must recognize that reactions occur as a result of the motion of electrons example, in the following reaction the curved arrows represent the motion, or flow, of electrons. this flow of electrons causes the chemical change shown: **general chemistry: an integrated approach - uh** - the use of half-reactions provides a general way to balance redox reactions. example: the reaction between permanganate ion (MnO_4^-) and oxalate ion ($\text{C}_2\text{O}_4^{2-}$) in acidic solution: when these ions combine in an acidic solution, they react to produce manganese ion (Mn^{2+}) and CO_2 gas. here is the unbalanced equation that describes this reaction ... **general chemistry chem 121 chem 141, ece 2018-2019 fall ...** - general chemistry: principles and modern applications, 9th edition lab-book: petrucci, harwood experiments in general chemistry, 7th edition available from: pearson education / İzmir tip kitapevi bornova objective of the course: to provide a thorough background of fundamental concepts in chemistry. **general chemistry - 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